

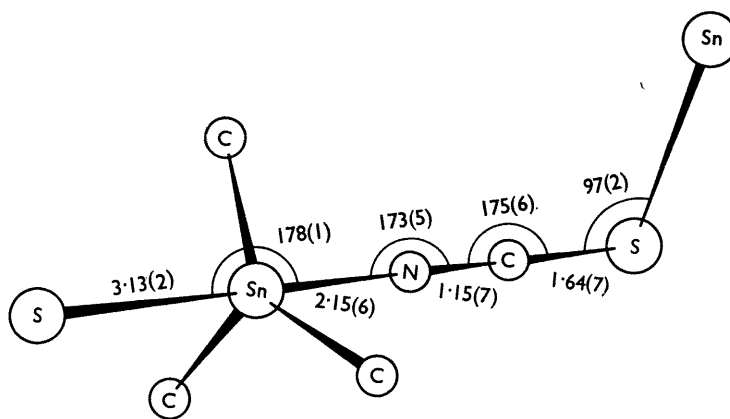
The Crystal Structure of Trimethyltin Isothiocyanate

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Summary The crystal structure of Me_3SnNCS consists of zig-zag $-\text{S} \dots \text{Sn}-\text{N}=\text{C}=\text{S} \dots \text{Sn}-$ chains, bent only at sulphur, with nearly planar trimethyltin groups.

MÖSSBAUER studies¹ of trialkyltin isothiocyanates indicate that the trialkyltin groups are planar, with five-co-ordinated tin; the infra-red spectra² are consistent with a

Weissenberg photographs taken with $\text{Cu-K}\alpha$ radiation. The structure was solved by the heavy atom method and has been refined by full-matrix least-squares with anisotropic temperature factors for the tin and sulphur atoms, and isotropic temperature factors for the carbon and nitrogen atoms, to a conventional 'R index' of 0.11 for 344 unique observed reflections.

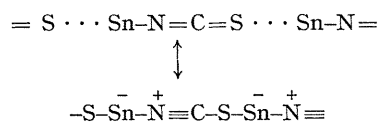


$\text{R}_3\text{Sn}-\text{N}=\text{C}=\text{S}$ rather than a $\text{R}_3\text{Sn}-\text{S}-\text{C}\equiv\text{N}$ structure, with a non-bonded $\text{S} \dots \text{Sn}$ interaction and a chain structure bent at nitrogen. In view of the ambiguities in the interpretation of the spectroscopic data, and also for comparison with the structure of silyl isothiocyanate,³ where linearity at nitrogen provides evidence of involvement of the silicon $3d$ orbitals in the bonding, we have determined the crystal structure of trimethyltin isothiocyanate by single-crystal X -ray diffraction.

The crystals are orthorhombic, $Pbca$; $a = 13.20 \pm 0.01$, $b = 10.28 \pm 0.01$, $c = 12.01 \pm 0.01$ Å, $Z = 8$, $D_c = 1.82$. Intensities were estimated visually from equi-inclination

The structure consists of zig-zag chains along the Z axis, all the atoms except those of the methyl groups lying approximately in planes perpendicular to the X -axis. The $\text{S}-\text{Sn}-\text{N}-\text{C}-\text{S}$ skeleton is linear within experimental error, the chains being bent at sulphur. The trimethyltin group is almost planar, with a mean $\text{N}-\text{Sn}-\text{C}$ angle of $95 \pm 2^\circ$, mean $\text{C}-\text{Sn}-\text{C}$ angle of $119 \pm 2^\circ$, and mean $\text{Sn}-\text{C}$ bond length of 2.13 ± 0.03 Å. The $\text{Sn}-\text{S}$ bond is appreciably longer than 2.47 Å found⁴ for the 'single' bond in $\text{Me}_3\text{Sn}-\text{SC}(\text{S})\text{NMe}_2$; the $\text{N}-\text{C}$ and $\text{C}-\text{S}$ distances are intermediate between the values in $\text{HN}=\text{C}=\text{S}$ (1.22 and 1.56 Å)⁵ and $\text{MeS}-\text{C}\equiv\text{N}$ (1.16 and 1.68 Å respectively).⁶

These results are consistent with a structure intermediate between the valence-bond extremes:



The angle at sulphur is very close to the value of 99.9° found in MeSCN. The stereochemical inactivity of the nitrogen lone-pair is accounted for by the above scheme, and does not imply 'p_π-d_π bonding' between nitrogen and tin.

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